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Corrosion is a galvanic process that can be prevented using cathodic protection. The deterioration of metals through oxidation is a galvanic process called corrosion. Protective coatings consist of a second metal that is more difficult to oxidize than the metal being protected.

17.5: Corrosion and Its Prevention - Chemistry LibreTexts

Prevention of Corrosion Preventing corrosion is of utmost importance in order to avoid huge losses. Majority of the structures we use are made out of metals. This includes bridges, automobiles, machinery, household goods like window grill, doors, railway lines, etc.

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To be an effective corrosion site, the crevice must be wide enough to allow the corroding electrolyte in and then provide 'stagnant' conditions. Crevice corrosion can therefore be a concern where gaps are a few micrometers wide, but there are no absolute or critical dimensions for crevices, below which

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corrosion is certain.

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